



## Experimental Apparatus

JASCO dual beam (Japan) UV-visible spectrophotometer model V-630, connected to an ACER compatible computer with spectra manager II software was used. The spectral slit width is 2 nm at speed up can be increased up to 8000 nm/min. All the measurements have been carried out in 1 cm quartz cell. The wavelength ranges were 200 - 400 nm at room temperature. Also, PASW statistics 18→ software program was used for statistical analysis.

## Materials and Reagents

### Pure standards

PAR and DCL were kindly provided by EIPICO(Egypt). Their purity was claimed to be as 99.50% and 99.80 % for PAR & DCL, respectively.

### Pharmaceutical formulations

Diclocin® tablets were purchased from the market (label claim: PAR 250 mg + DCL 50 mg) produced by Cipcopharmaceuticals, India.

### Solvents

HPLC grade Methanol was purchased from LiChrosolv, Merck KGaA (Germany). All of measurements have been accomplished by using 90% Methanol.

### Standard solutions

Standard stock solutions (1 mg/mL) of PAR and DCL were prepared in 90% methanol. Working standard solution of PAR (40

µg/mL) and DCL (50 µg/mL) were prepared by further dilution with 90% methanol.

### Laboratory prepared mixtures

Different ratios of PAR & DCL were performed by transferring aliquots from their standard solutions to volumetric flasks (10 mL) and then dilution was carried out with 90% methanol.

## Procedure

### Construction of calibration curves

For PAR: Working solutions equivalent to 4-22 µg/mL were prepared by addition of aliquots (1, 1.50, 2, 2.50, 3, 3.50, 4, 4.50, 5, 5.50 mL) of PAR working standard solution (40 µg/mL) to 10 mL volumetric flasks followed by dilution with 90% methanol. For DCL: Working solutions equivalent to 5-45 µg/mL were prepared by adding aliquots (1, 1.50, 2, 2.50, 3, 3.50, 4, 4.50, 5, 6, 7 mL) of DCL working standard solution (50 µg/mL) to 10 mL volumetric flasks followed by dilution with 90% methanol. Measurements of the absorption spectra were carried out at room temperature over the wavelengths (200-400 nm).

### For Spectrum Subtraction method

The method relies on subtracting the spectrum of Y from the spectrum of the mixture (X + Y), therefore we can obtain the zero-absorption spectrum of X again. This can be summarized as the following:

$$(X + Y) - Y = X$$

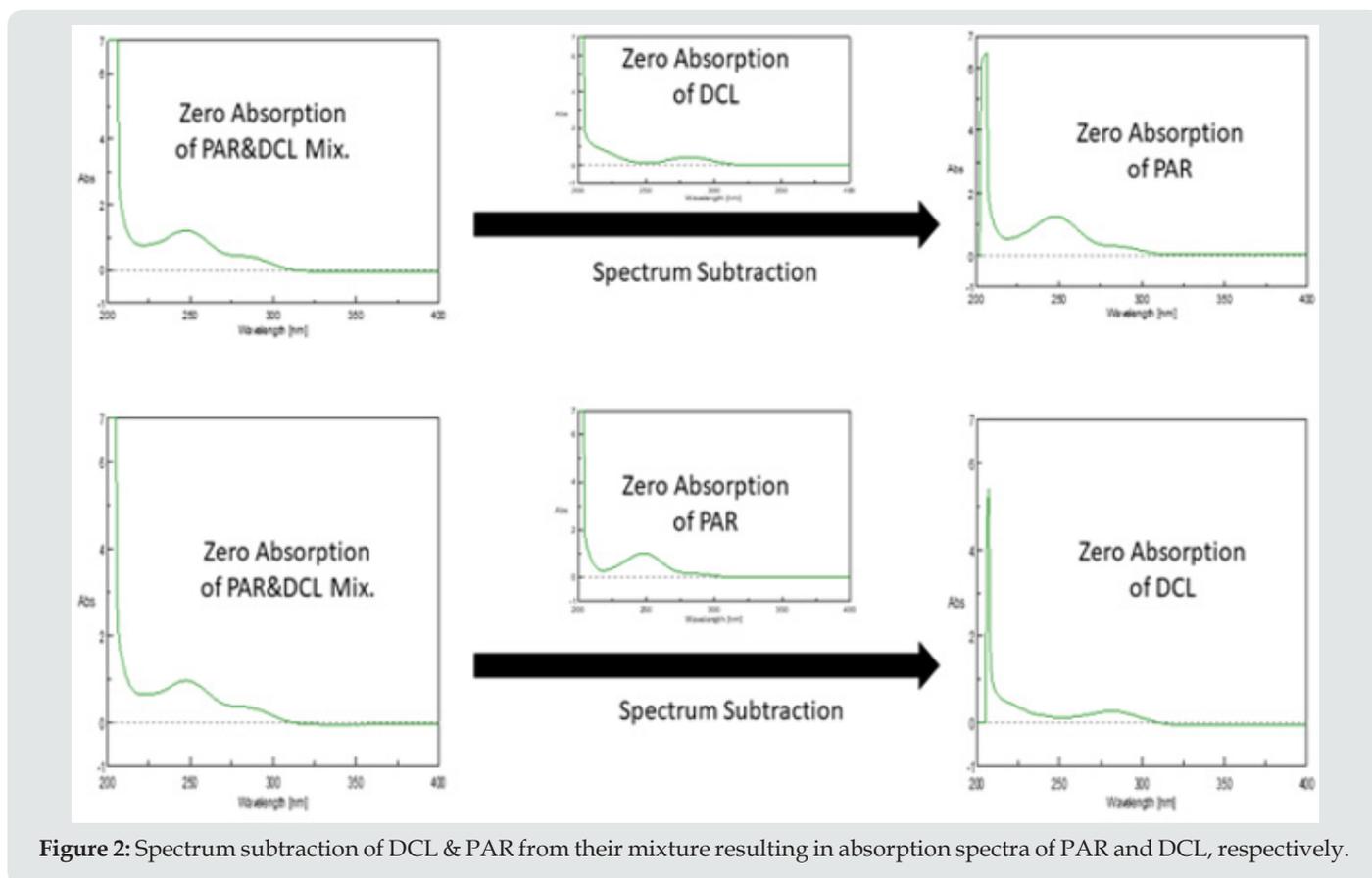


Figure 2: Spectrum subtraction of DCL & PAR from their mixture resulting in absorption spectra of PAR and DCL, respectively.

The concentration of X is calculated from the corresponding regression equation obtained by plotting the absorbance values of the zero order absorption spectra of X at its  $\lambda_{\max}$  against the corresponding concentrations. Zero absorption spectra of PAR

& CAF can be recovered from their mixture through spectrum subtraction of CAF and PAR, respectively (Figure 2). Zero absorption spectra of CAF and PAR are shown in (Figure 3).

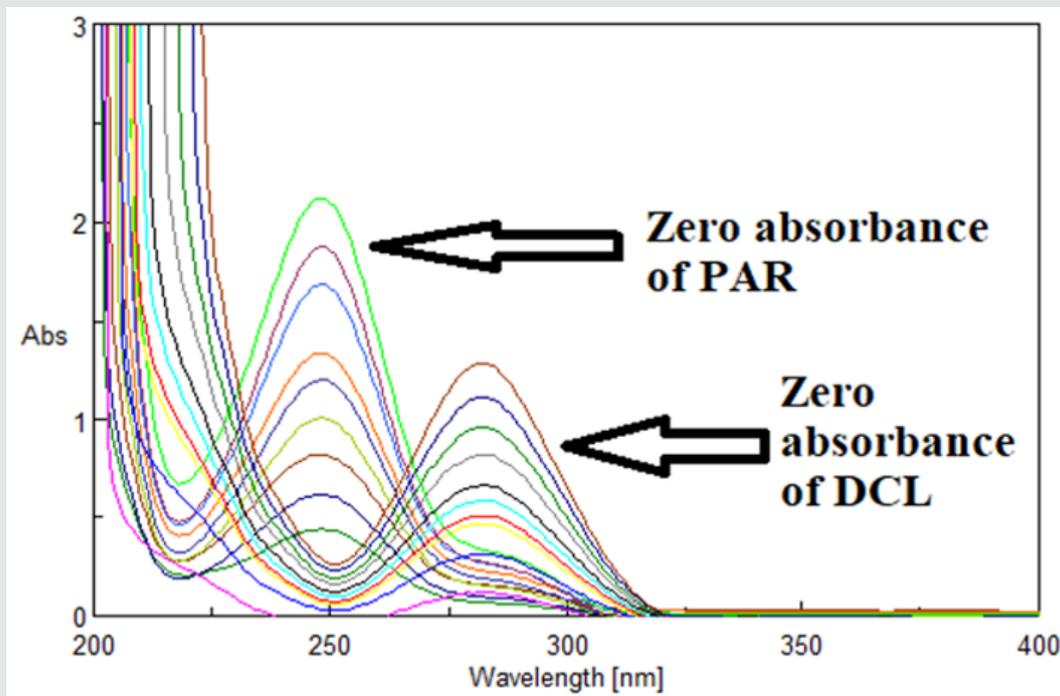


Figure 3: Zero absorption spectra of PAR overlaid with zero absorption spectra of DCL.

### Analysis of laboratory prepared mixtures

The spectra of the mixtures were measured after preparation of different ratios of the laboratory prepared mixtures then handled in the same conditions as described under each method.

### Application to pharmaceutical formulation

10 tablets of Diclofenac<sup>®</sup> were weighed and crushed then an amount equivalent to 50 mg PAR and 10 mg DCL in each tablet was transferred into a volumetric flask (50 mL) and diluted with 90% methanol as follow: First, 30 mL of 90% methanol were added and sonicated then dilution was carried out to the mark and filtered. Second, 10 mL of the dilution was transferred into a 100 mL volumetric flask to give a concentration equivalent to 100  $\mu\text{g}/\text{mL}$  PAR and 20  $\mu\text{g}/\text{mL}$  DCL. Third, any further dilutions were carried out in volumetric flasks (10 mL) and treated in the same way as described under each method.

## Results and discussion

### Method Optimization

Two major problems were found during the analysis of PAR & DCL binary mixture; first, the overlapped spectra between the absorptivity of both drugs, and second, PAR, the main (major) constituent, had unfortunately very high absorbance, while DCL, the minor component, had low absorbance value. Intrinsically, sample enrichment technique [20] was used in which the concentration of DCL (the minor component) in their dual mixtures was increased to

facilitate its determination. This was carried out by adding a fixed amount of standard DCL to each experiment when combined with PAR, then subtraction of its concentration before the calculation of the required concentration of DCL. Sample enrichment technique has been used for solving the same problem in the analysis of other drug mixtures of different drug ratios [21,22].

### Spectrum Subtraction method

248 and 283 nm absorbances were used for determination of PAR & DCL in presence of each other, respectively. The calibration curves revealed accepted linear relationships between concentrations and absorbance in a range of 4-22  $\mu\text{g}/\text{mL}$  for PAR and 7.5-45  $\mu\text{g}/\text{mL}$  for DCL with correlation coefficients of  $\geq 0.9990$  for both drugs. The accuracy of the method illustrated accepted values with  $99.94\% \pm 0.98$  for PAR and  $99.85\% \pm 0.55$  for DCL. The specificity of the method demonstrated accepted values with  $99.97\% \pm 1.51$  for PAR and  $100.86\% \pm 1.36$  for DCL. The results are detailed in Table 1.

Spectrum subtraction is very easy and simple as it depends on zero absorption spectra without the need of extra processing. It is having few steps to get the zero order spectra of the desired drug, but it suffers from noise interference while acquiring the desired drug concentration by subtraction.

### Method validation

The method was validated according to ICH guidelines [17]. The linear regression data for the calibration curve showed good linear

relationship. (Table 1). The accuracy was calculated by analyzing the standard addition where satisfactory results were obtained as shown in Table 1. The specificity of the method was calculated by assaying the laboratory prepared mixtures of PAR & DCL within

the linearity range and good results were obtained (Table 1). The intra- and inter-day precisions were calculated by the analysis of 3 different concentrations of the drugs 3 times on the same day and on 3 successive days (Table 1).

**Table 1:** Assay parameters and validation results obtained by applying Spectrum subtraction method.

Mixture	PAR & DCL	
	DCL	PAR
Method Parameters		
Wavelength (nm)	283	248
Linearity range ( $\mu\text{g}/\text{mL}$ ) (n=3)	7.5-45	22-Apr
Intercept	-0.0905	0.08
Slope	0.0303	0.0911
Correlation coefficient (r)	0.9991	0.999
Accuracy (Mean $\pm$ SD)	99.85 $\pm$ 0.55	99.94 $\pm$ 0.98
<b>Precision (<math>\pm\%</math>RSD)</b>		
Repeatability	98.72 $\pm$ 0.18	99.11 $\pm$ 0.12
Intermediate precision	98.73 $\pm$ 0.14	99.81 $\pm$ 0.48
Specificity (Mean $\pm$ SD)	100.86 $\pm$ 1.36	99.97 $\pm$ 1.51

### Application to Pharmaceutical Formulation

The proposed method was successfully applied for determination of PAR and DCL in their pharmaceutical formulation

(Diclofenac Sodium plus  $\rightarrow$  tablets). The results were acceptable and with sufficient agreement with the labeled amounts. The standard addition technique was applied and showed that no interference of the excipients was observed (Table 2).

**Table 2:** Analysis of the pharmaceutical preparation (Dicloclin<sup>®</sup> tablets) by applying proposed method.

	Spectrum subtraction							
	DCL				PAR			
			Recovery%				Recovery%	
	Tablet	Standard Added	Tablet	Added	Tablet	Standard Added	Tablet	Added
	Taken ( $\mu\text{g}/\text{mL}$ )	( $\mu\text{g}/\text{mL}$ )			Taken ( $\mu\text{g}/\text{mL}$ )	( $\mu\text{g}/\text{mL}$ )		
	2	8	101.91	100.04	10	9	101.29	101
		9.5	101.77	100.28		10	101.58	99.06
		10	99.61	99.24		11	101.46	99.76
Mean			101.1	99.85			101.45	99.94
SD			1.29	0.55			0.15	0.98

### Statistical Analysis

Statistical comparison of the proposed method was performed through One-way ANOVA method by using PASW statistics 18  $\rightarrow$

software program in which there was no significant difference between the proposed method and the reference one [4] as shown in Table 3.

**Table 3:** Statistical comparison of the results obtained by the proposed method and the reference method using One-way.

Tablets	Drugs		Sum of Squares	df	Mean Square	F	Sig.
Dicloclin <sup>®</sup> tablets	PAR	Between Groups	3.45	1	3.45	2.916	0.163
		Within Groups	4.734	4	1.183		
		Total	8.184	5			
	DCL	Between Groups	1.55	1	1.55	0.786	0.425
		Within Groups	7.886	4	1.971		
		Total	9.436	5			

## Conclusion

Spectrum subtraction method was successfully applied for the determination of paracetamol and Diclofenac Sodium in their binary mixtures and in their dosage form. The proposed method is simple, sensitive and accurate and could be used for routine analysis by using simple technology or instruments. By comparison with the previous reported methods, it was concluded that spectrum subtraction method is very simple and doesn't require extra processing. Statistical comparison revealed that there was no observed significant difference between the proposed method and the reference one.

## References

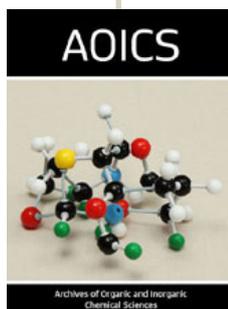
1. Yehia AM, Abd El-Rahman MK (2015) Application of normalized spectra in resolving a challenging Orphenadrine and Paracetamol binary mixture. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy*, Elsevier BV 138: 21-30.
2. Barry M, Mahood AJ, Hamezh M (2009) Spectrophotometric Determination of Diclofenac sodium in Pharmaceutical preparations. *Journal of Kerbala University* 7(2): 310-316.
3. Scholer DW, Boettcher I, Ku EC, Schweizer A (1985) Pharmacology of diclofenac sodium (Voltaren®). *Seminars in Arthritis and Rheumatism*, Grune & Stratton Inc 15(2): 61-64.
4. Brogden, RN, Heel RC, Pakes, GE, Speight TM, Avery GS (1980) Diclofenac Sodium: A Review of its Pharmacological Properties and Therapeutic Use in Rheumatic Diseases and Pain of Varying Origin. *Drugs* 20(1): 24-48.
5. Saheb DJ, Reddy NR, Chakravarthy IE (2004) Simultaneous determination of paracetamol and diclofenac sodium from combined dosage forms by absorbance difference method. *Asian Journal of Chemistry* 16(2): 767-772.
6. Hegazy MA, Elshahed MS, Toubar SS, Helmy MI (2018) Efficient processing of Single and Multiple Spectral Variables for Resolution and Quantitation of Paracetamol, Chlorzoxazone and Diclofenac. *Journal of Advanced Pharmacy Research* 2(4): 269-282.
7. Phaneendra D, Nagamalleswar G (2012) Quantitative analysis of paracetamol and diclofenac in combined dosage form by first derivative and simultaneous equation method in application to the determination of dissolution study. *International Journal of Pharmaceutical Sciences and Research* 3(10): 3871-3876.
8. Sharma R, Pathodiya, G, Mishra GP, Sainy J (2010) Spectrophotometric Methods for Simultaneous Estimation of Paracetamol and Diclofenac Sodium in Combined Dosage Form by Application of Hydrotropic Solubilization. *Journal of Pharmaceutical Sciences and Research* 2(12): 821-826.
9. Sebaiy, MM, El-adl SM, Mattar AA (2020) *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy* Different techniques for overlapped UV spectra resolution of some co-administered drugs with paracetamol in their combined pharmaceutical dosage forms. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy* Elsevier BV 224: 117429.
10. Walash MI, Ibrahim F, Abo S, Abass E (2017) Development and Validation of HPLC Method for Simultaneous Estimation of Famotidine, Paracetamol and Diclofenac in their Raw Materials and Pharmaceutical Formulation. *Analytical Chemistry Letters* 7: 421-437.
11. Jana K, Adhikari L, Moitra S, Behera A (2011) Analysis of multicomponent drug formulations (diclofenac and paracetamol). *Asian Journal of Pharmaceutical and Clinical Research* 4: 2-4.
12. Badgujar MA, Pingale SG, Mangaonkar KV (2011) Simultaneous Determination of Paracetamol, Chlorzoxazone and Diclofenac Sodium in Tablet Dosage Form by High Performance Liquid Chromatography. *E-Journal of Chemistry* 8(3): 1206-1211.
13. Nayak VG, Bhate VR, Purandare SM, Dikshit PM, Dhupal SN, et al. (1992) Rapid Liquid Chromatographic determination of Paracetamol and Diclofenac Sodium from a combined pharmaceutical dosage. *Drug Development and Industrial Pharmacy* 18(3): 369-374.
14. Dighe VV, Sane RT, Menon SN, Tambe HN, Pillai S, et al. (2006) Simultaneous Determination of Diclofenac Sodium and Paracetamol in a Pharmaceutical Preparation and in Bulk Drug Powder by High-Performance Thin-Layer Chromatography. *Journal of Planar Chromatography* 19(112): 443-448.
15. Mohammad ALI, Sharma S (2009) Separation of Co-existing Paracetamol and Diclofenac Sodium on Silica Gel "H" Layers using surfactant mediated mobile phases: Identification of Diclofenac Sodium from human urine. *Farmacia* 57: 201-11.
16. Solangi A, Memon S, Mallah A, Memon N (2010) Determination of ceftriaxone, ceftizoxime, paracetamol, and diclofenac sodium by capillary zone electrophoresis in pharmaceutical formulations and in human blood. *Turk J Chem* 34(6): 921-933.
17. Transactions ECS, Society TE (2018) Voltammetric Determination of Diclofenac in the Presence of Paracetamol and Naproxen by an Artificial Neural Network Model Using a Carbon Paste Electrode G. Y. Aguilar-Lira. *ECS Transactions* 84(1): 195-205.
18. Okoth OK, Yan K, Liu L, Zhang J (2015) Simultaneous Electrochemical Determination of Paracetamol and Diclofenac Based on Poly (diallyl dimethylammonium chloride) Functionalized Graphene. *Electroanalysis* 28(1): 76-82.
19. Afshar Esmail, Jalali F (2016) sensitive simultaneous determination of paracetamol and diclofenac based on au nanoparticles - functionalized graphene/poly (l-arginine) glassy carbon electrode. *Journal of the Chilean Chemical Society* 61(1): 2846-2851.
20. Lotfy HM, Tawakkol SM, Fahmy NM, Shehata MA (2014) Successive spectrophotometric resolution as a novel technique for the analysis of ternary mixtures of pharmaceuticals. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy* Elsevier BV 121(5): 313-323.
21. Moussa BA, Mahrouse MA, Fawzy MG (2018) Different resolution techniques for management of overlapped spectra: Application for the determination of novel co-formulated hypoglycemic drugs in their combined pharmaceutical dosage form. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy*, Elsevier BV 205: 235-242.
22. Lotfy HM, Mohamed D, Mowaka S (2015) A comparative study of smart spectrophotometric methods for simultaneous determination of sitagliptin phosphate and metformin hydrochloride in their binary mixture. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy* Elsevier BV 149: 441-451.
23. ICH (2005) Validation of Analytical Procedure: Text and Methodology Q2 (R1). ICH Steering Committee 1994.



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