Determination of Malathion in Water Using Liquid Chromatography & Mass Spectroscopy

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Received: January 25, 2018; Published: February 05, 2018

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Editorial

One of the most common wastewater pollutants are organo-phosphorus pesticides, which have a complex of properties - high solubility, toxicity, mobility in the environment. In recent years, this method has become an innovative analytical technology for the most part used in medical practice and in pharmaceuticals. With development, the method begins to acquire a practical application in other branches of human life activity [1,2]. In this paper, the results of studies on the detection of Malathion (MLT)-O, O-Dimethyl-S- (1,2-dicarbethoxyethyl) dithiophosphate in aqueous media. The purpose of this study was to develop liquid chromatography & mass spectroscopy method for the determination of Malathion in water solution. This method will provide experimental results from testing conducted to evaluate sensitivity, selectivity, linearity, precision, accuracy, dilution integrity and various additional stability tests.

Reagents and Equipment

The Experiments were carried out using HPLC Grade Methanol, Acetonitrile and Deionized water supplied by Carl Roth, Germany. Formic Acid (98% ACS grade) Carl Roth Germany and Malathion (500 g/L). The research was carried out using Shimadzu's Nexera X2 high-performance liquid-Chromatography equipped with CBM-20A controller, two Binary LC-30AD Pumps, connected with DGU-20A5R degasser unit, a SIL-30ACMP auto sampler and column temperature unit CTO-30A (Sciex, Germany). X BridgeC8 1.0 mm x 100 mm dimensions with 3.5µm particle sizes and 136 Å pore size column was used (Waters, USA). The LC was coupled with Sciex's Triple Quad™ 4500 Mass spectrometry by the Turbo V™ Ion Source (Sciex, USA). Data processing was performed using the software Analyst 1.6.3 (Sciex, Germany).

Conditions of Mass Spectrometry

The tuning of MS parameters was conducted in positive ionization mode using Sciex's Triple Quad™ 4500 MS for Malathion (MLT). MLT was injected in the MS in 1ng/mL concentration diluted

in 1:1 methanol: 0.1% Formic Acid and a flow rate 7µL/min. and gave predominant singly charged protonated precursor $[M+H]^+$ ions at m/z of 331.1 Da in Q1 full scan spectra. In the product ion mass spectrum of MLT the most abundant and consistent ions were observed at m/z 127.1 Da.

Preparation of Standards Working Solutions

The concentration of the stock solution was 0.5 mg/mL. Two secondary stock solution 5.0 and 1.0µg/mL were prepared from the stock solution. Serial dilution were made to obtain a final concentration of nine Calibration Curve Standards Working Solutions (CCS WS) as 1.0, 2.0, 5.0, 10.0 20.0 50.0, 100.0, 200.0 and 500.0ng/mL respectively. Four levels of Quality Control Working Solution (QC WS) preparation, separate secondary stocks were prepared and QC WS were diluted to had the concentration 1.0, 3.0, 30.0 and 150.0for Lower Limit of Quantization QC (LLOQ QC), Low Quality Control (LQC), Medium Quality Control (MQC) and High-Quality Control (HQC) respectively. Internal Standard (IS) was chosen to be Solifenacin Succinate, one of the laboratory standards, and the Working Solution of the IS was prepared to have the concentration of 20ng/mL.

Sample Preparation

Sample Preparation was a one-step dilutes and shoot. Into 900µL Deionized Water, 50µL of IS and 50µL of each CCS/QC was pipette into each sample. To be diluted twenty times than the working Solutions concentration. Those samples were prepared in 1.8mL glass vials and after mixing they had placed in the auto sampler of the LC. The final concentrations were as the following for Calibration Curve Standards (CCS) 0.05, 0.1, 0.25, 0.50, 1.00, 2.50, 5.00 and 10.00ng/mL, and for Quality Control Samples (QC) 0.05, 0.15, 1.50 and 7.5ng/mL respectively.

Method Performance Check Results

For checking the performance of the method, precision and accuracy tests were conducted. Precision of an assay is defined as

the closeness of the results when replicate assays of a test sample are carried out. The precision is expressed as %CV. The accuracy of a method is defined as the proximity of the determined values to actual concentrations of the samples. The accuracy is expressed as % Accuracy. Both the inter-run and intra-run precision and accuracy were assessed by replicate analysis of samples containing known amounts of the Analyte (QCs). The results for inter-run and intra-run precision and accuracy tests of four replicate QC samples at LLOQ QC, LQC, MQC and HQC concentration levels are listed in (Tables 1 & 2). Assay sensitivity was assessed by precision (%CV), accuracy (%Nominal) and signal to noise (S/N) ratio at CCS1 concentration of all Calibration Curves. The results are listed in (Table 3).

Table 1: Accuracy Results.

QC Level of Malathion	% Accuracy (n = 4)				
	I	II	III	IV	Overall
LLOQ QC (0.05 ng/mL)	104.5	98.7	110.6	103.3	104.3

Table 3: Sensitivity Results.

n = 4	PrAcc I	PrAcc II	PrAcc III	PrAcc IV	Mean	STD	%CV	%Accuracy
Concentration	0.053	0.052	0.052	0.053	0.052	0.001	1.4	104.8
Signal/Noise	94.6	84.8	62.0	71.7				

Conclusion

The developed method was proved to be fast and efficient, more over this method showed to be simple, rugged, sensitive and selective for the quantitative determination of Malathion in aqueous solution using Liquid Chromatography tandem Mass Spectrometry. Thus, this method can be used for widespread environmental analysis for determination of pesticides in water.

Acknowledgment

Darmantest Laboratories are acknowledged for the support and the help they provided to develop and process this method.

LQC (0.15 ng/mL)	99.4	97.6	99.6	98.0	98.7
MQC (2.00 ng/mL)	110.7	103.9	101.6	103.7	105.0
HQC (7.5 ng/mL)	112.4	106.1	102.0	106.5	106.7

Table 2: Precision Results.

QC Level of Malathion	% CV (n = 4)				
	I	II	III	IV	Overall
LLOQ QC (0.05 ng/mL)	8.9	5.8	6.8	6.2	4.7
LQC (0.15 ng/mL)	2.8	2.2	3.9	1.2	1.0
MQC (2.00 ng/mL)	4.2	2.6	0.5	1.4	3.8
HQC (7.5 ng/mL)	2.7	3.0	2.9	2.5	4.0

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DOI: [10.32474/AOICS.2018.01.000112](https://doi.org/10.32474/AOICS.2018.01.000112)

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